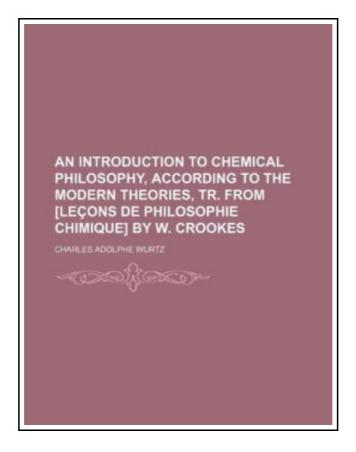
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English. Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1867 Excerpt: . Formiate of propyl. Butyric acid. Here is one of the typical equations, --Wge + Ge =h6 +B Acetic acid. Alcohol. Acetic ether. Is it possible to express in a clearer and more simple manner this fact, --that the reaction of acetic acid on alcohol consists of an exchange of elements, and that the formation of acetic ether is necessarily connected with that of water? Certainly the typical equation gives account of the essential conditions, and, in a manner, of the mechanism of the reaction. There is a singular difference of opinion among some chemists on this subject. Among the detractors of the typical notation, some affirm that it says too much, others regret that it does not say enough. Admitting, say the former, these exchanges of simple bodies for groups in the systems which it considers as typical, the theory implies hypotheses on the molecular grouping. It does not confine itself to representing facts, it goes beyond them. It is true, say the others, that it perfectly interprets certain reactions, but it is powerless to express them all. For after all, these molecular changes, these double decompositions which it depicts so well, are not the only reactions; there are molecular additions and subtractions; and when it becomes necessary to account for those more or less profound changes, which attack not only the external scaffolding, but the very substance of the molecule, the typical formulso afford very slight, if any, assistance. The following considerations will reduce...

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